

Lab session 2: Dye solar cells

Objective:

Fabricate and characterize a Dye solar cell.

Materials

Nanoporous TiO₂ Films
 3-methoxypropionitrile, (3-MPN)
 Iodine (I₂)
 Lithium Iodide (LiI)
 1-methylbenzimidazole (MBI)
 cis-(NCS)₂(2,2'-bipyridyl-4,4'-dicarboxylate)₂Ru(II) (N3 ruthenium dye)
 acetonitrile and tert-butyl alcohol (1:1 solution)
 Platinized FTO Glass
 clamps
 Lamp
 2 Polimeters
 Variable resistor
 Power meter

Procedure:

- DSC fabrication

Immerse the TiO₂ electrode into a 0.5mM N3 dye solution in a mixture of acetonitrile and tert-butyl alcohol (volume ratio: 1:1) and kept at room temperature for 20–24 h to complete the sensitisier uptake
 Assemble the dye-covered TiO₂ electrode and the Platinised glass in a sandwich cell and fix it with clamps.

- DSC characterization

With the powermeter, measure the light intensity of the lamp.

Shine light over the samples and with the polimeters measure the Current and Voltage of the cell (i-V curve) by fixing the potential through the variable resistor from short circuit current (i_{sc}) to open circuit potential (V_{oc}). Take 10 values of i-V. Measure the surface (S) of the TiO₂ film and normalize the current to obtain current density ($J=i/S$).

Results

Make the plots of J-V curve and Power-V and provide the values of J_{sc} , V_{oc} , FF and Efficiency.

References: ¹⁻³

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3. Fabregat-Santiago, F.; Bisquert, J.; Garcia-Belmonte, G.; Boschloo, G.; Hagfeldt, A. Influence of electrolyte in transport and recombination in dye-sensitized solar cells studied by impedance spectroscopy. *Solar Energy Materials and Solar Cells* **2005**, *87* (1-4), 117-131.